## PATENT SPECIFICATION

NO DRAWINGS



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## COMPLETE SPECIFICATION

## Methods for Dyeing Hair and like Keratinous Materials

We, Societe Monsavon-L'OREAL, a French body corporate, of 14, rue Royale, Paris 8éme, (Seine), France, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:-

This invention relates to methods for dye-10 ing hair, particularly human hair, and like

keratinous materials.

In our Application No. 22892/56 (Serial No. 797,174) we have described a process for dyeing human hair and other keratinous fibres 15 which consists essentially in (a) preparing an aqueous solution adjusted to an alkaline pH, 5:6-dihydroxy-indole ammonia, an amine, or an ammonium salt (b) applying this solution to the hair or similar material to be dyed and rinsing the latter after a period of contact varying in accordance with the desired shade. In this way, relatively light shades ranging from light ash-blond to dark ash-blond are obtained for a standard 25 period of contact of the order from 20 to 30 minutes at room temperature, which period may be reduced by slight heating, for example, in the case of hair-dyeing, by means of a dry heat hood or a steam hood. It is furthermore possible, in accordance with the method of the aforesaid application, to obtain a more rapid development of the shade by adding to the dyeing solution, at the time of application, an oxidising agent, such as sodium bromate, a 35 persalt such as sodium perborate or ammonium persulphate, or simply hydrogen peroxide. There is to be noted that where an oxidising agent is used it is necessary to make a wellbalanced choice of the quantity added, a minimum quantity being necessary for a sufficiently rapid formation of the coloured pigments while an excess of this agent has the effect of more or less destroying said coloured pigements intermediately formed and thus 45 of reducing the depth of the final shade.

As indicated in said application, it is pos-

sible to obtain dark shades closely comparable to the natural shades of the hair by operating in two steps, i.e. by first treating the hair with an alkaline solution of 5:6-dihydroxyindole at room temperature for 15 to 30 minutes and then, after rinsing with water, applying to the hair a solution of an oxidation catalyst.

It is an object of the present invention to provide a process for the dyeing of human hair and like keratinous fibres by means of 5:6dihydroxy-indole while still operating in two steps, wherein the 5:6-dihydroxy-indole solution applied in the first step, instead of being

alkaline is either acid or neutral.

According to the present invention a process for dyeing live hair and like keratinous materials consists essentially in impregnating the hair or like keratinous material at room temperature with an aqueous solution of 5:6dihydroxy-indole having a pH-value of at most 7, allowing the impregnation to proceed for about 5 to 60 minutes, blotting off the excess solution, and developing the shade in situ by applying an aqueous solution capable of inducing oxidation of 5:6-dihydroxy-indole.

The pH-value of the aqueous 5:6-dihydroxy-indole solution to be applied in a first step for carrying out the invention may be obtained in various ways. For example, there may be employed an aqueous solution only containing the 5:6-dihydroxy-indole, with the exclusion of any other reactant, the pH-value of such a solution being of course slightly acid. It is also possible to acidify this aqueous solution more or less strongly, for example by means of an inorganic acid, an organic acid, e.g. formic, acetic or monochloroacetic acid, or any compound containing at least one acid 85 hydrogen atom.

The concentration of 5:6-dihydroxy-indole in the solution as applied to the hair or other keratinous material may vary within fairly wide limits. Generally it will be between 0.1% and 10% and preferably between 1% and 2%, by weight of the solution. The term

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"aqueous solution" as employed here means any solution the principal solvent of which is water. It is of course possible, without departing from the scope of the invention, to use in the preparation of such a solution water alone or in admixture with the water-miscible solvents, notably mixtures of water and ethyl alcohol or isopropyl alcohol or water and gly-

cols or glycol ethers.

The aqueous solution of 5:6-dihydroxyindole to be applied first may be prepared at the time of use. However, 5:6-dihydroxyindole, like many other organic compounds, is relatively much more stable in acid or neutral medium than in alkaline medium and it is possible to have it prepared in advance in the form of any aqueous solution and, whenever desired, further to improve the stability of such a solution by adding thereto a small amount of a reducing agent, such as, for example, an alkali metal sulphite, bisulphite of hydrosulphite, this addition preferably being made first, i.e. before the 5:6-dihydroxyindole is added. The stability of aqueous solutions of 5:6-dihvdroxy-indole may also be enhanced by adding thereto a small quantity preferably 0.2% to 0.5% by weight, of a sequestering agent, i.e. a compound capable of forming complexes with ions of alkalineearth metals or of heavy metals, often existing as trace impurities. The sequestering agent may be, for example, ethylenediamine-tetracetic acid or one of its salts, such as that known under the Registered Trade Mark "Trilon" B.

When the aqueous 5:6-dihydroxy-indole solution has been in contact with the hair or other material for a period of 5 to 60 minutes the excess solution is blotted off, for example

40 by pressing the hair in a towel.

The shade is then developed in situ by applying to the hair or other keratinous material an aqueous solution capable of inducing oxidation of the 5:6-dihydroxy-indole.

The choice of this oxidation inducing solution, as also the length of time during which it should be maintained in contact with the hair or the other keratinous material to be dyed, depends to some extent upon the final shade it is desired to obtain. It is also possible to predetermine the final shade by adjusting the pH of the developing solution to a preselected value, e.g. by including acid or alkaline materials therein.

In one method of carrying out the process of the invention there is used as oxidising agent the oxygen of the ambient air, if desired ozonised. In such an embodiment the development of the shade, which amounts to the rapid fixing of the atmospheric oxygen by the hair or similar material impregnated with 5:6-dihydroxy-indole, is effected by applying in a second step as a developing medium an aqueous solution containing an alkaline sub-

stance consisting preferably of ammonia, a 65 basic ammonium saft such as ammonium carbonate or triammonium phosphate, or an organic amine. It is generally advantageous also to incorporate in this solution an oxidation catalyst which accelerates the process of fixing the oxygen of the ambient air, for example small quantities of a cobalt or manganese, salt, such as the sulphate or the lactate. The aqueous solution containing the alkaline substance and, if desired, the oxidation catalyst is left in contact with the hair or similar material for 10 to 60 minutes as necessary for the penetration of the reactants. The hair is then rinsed clear with water and dried. The development of the shade, which is initiated by the alkalisation, may proceed for some time, this process then constituting

a progressive dyeing process.

In a second method of carying out the invention, there is applied in a second step, for the development of the shade, an aqueous solution of an oxidizing agent, such as, for example, hydrogen peroxide, or an alkali metal or ammonium iodate, periodate, or persulphate or an alkali metal salt of an N-chlorinated derivative of a sulphonamide. This solution is allowed to act for a period of time sufficient for the development of the shade, generally from 5 to 20 minutes, and the hair is then rinsed, shampooed, rinsed again and dried. The pH of this aqueous oxidising solution, applied in the second place, may be adjusted to any desired value, either acid, neutral or alkaline, by the addition of any appropriate reactant, for example an inorganic 100 or organic acid, a salt having an acid reaction, a buffer, a salt having a basic reaction or a base such as ammonia or an organic amine. The choice of the most suitable value depends mainly, but not solely, upon the nature of the 105 oxidising agent employed. If the latter is hydrogen peroxide, an alkali metal or ammonium iodate or periodate or an alkali metal salt of an N-chlorinated derivative of a sulphonamide, it is generally preferably to operate in a basic medium, whereas the operation is preferably carried out in an acid medium if the oxidising agent is an alkali metal or ammonium persulphate. However, if the oxidising agent employed is an iodate or a perio- 115 date, it is preferable not to operate at a pHvalue lower than 3 and to avoid the use of a strong inorganic acid for adjusting the pH to the desired value.

Whatever the pH chosen for the develop- 120 ment of the shade, the concentration of the oxidising agent in the aqueous solution may be chosen between fairly wide limits, depending upon the desired shade and upon the own solubility of the specific oxidising agent which 125 is to be used. Generally speaking, the quantity of oxidising agent to be employed corresponds to 0.5 to 10 g. per 100 cc. of solution, the volume of oxidising solution to be

employed being substantially the same as the Example II quantity of dihydroxyindole solution. An impregnating solution as set forth in When the oxidising solution employed in Example I is applied to naturally white the second step is acid or neutral, i.e. free from This solution is left in contact with the hair 70 ammonia and organic amines, it is probable that the pigment formed results from the for 20 minutes, then blotted off, and the folformation of the quinone corresponding to 5, lowing solution is then applied: -6-dihydroxyindole. However analytical data has shown that if the oxidising solutions are Sodium chloride -Hydrogen peroxide (20 vol.)  $(6\% \text{ of } \hat{H}_2O_2)$  - - - - Water to make - - - -10 rendered alkaline by means of ammonia or an 2.5 cc. amine, the molecule of the pigment obtained - 100 cc. contains nitrogen issuing from the ammonia This solution is left in contact with the hair for 1 hour, and the hair is then rinsed. or the amine employed. The same holds true shampooed, rinsed again and dried. A light when the oxidation is effected through the intermediary of ambient air and in the pregrey shade is obtained, which remains subsence of ammonia or an amine, according to stantially unchanged in the following weeks. the first embodiment of the invention. According to an accessory feature of the EXAMPLE III. invention and whatever the specific embodi-Almost white live hair is impregnated with ment to be dealt with, there can be advanan impregnating solution as set forth in tageously added to the developing solution a Example I. After contact for 20 minutes, the quantity of a neutral, strongly ionised inexexcess solution is blotted off and a second solution having the following composition is pensive electrolyte such as, for example, sodium chloride or sodium sulphate. then applied:-The dyeing process according to the inven-Sodium chloride -Water - - - - - 20% ammonia - - -100 cc. 90 tion may be carried out, not only by means of aqueous solutions as referred to above, 20% ammonia 17 cc. Hydrogen peroxide (20 vol.) (6% of H<sub>2</sub>O<sub>2</sub>) - - but also by means of aqueous creams, pastes or jellies, the adjustment to the required consistency being effected by any method usual This second solution is left in contact with in hair dyeing. the hair for 30 minutes, and the hair is then rinsed shampooed, rinsed again and dried. The The use of 5,6-dihydroxy-indole, under the above-defined conditions, constituting the hair is dyed jet-black. object of the present invention may be combined with that of di- or tri-hydric phenois EXAMPLE IV. capable of dyeing hair and other keratinous Almost white live hair is impregnated with an impregnating solution as set forth in 100 fibres at room temperature. The following examples will serve to illus-Example I. After contact for 20 minutes, the excess trate the invention:-Example I illustrates the case where the solution is blotted off and the following soludevelopment is effected in a basic medium and tion is then applied:— 105 Sodium chloride by oxidation in the ambient air, while the other examples illustrate cases where the Monomethylamine (in 33% aqueous solution) development is effected by means of an oxi-10 cc. Hydrogen peroxide (20 vol.) dising solution.  $(6^{0/}_{/0} \text{ of } \hat{\mathbf{H}}_{2}\mathbf{O}_{2})$  - -EXAMPLE I. 2.5 cc. An impregnating solution having the follow-Water to make -100 cc. This solution is left in contact with the ing composition is prepared:-5,6-dihydroxy-indole - hair for 20 minutes, and the hair is then Acetic acid - - - - - Water to make - - -5 cc. rinsed, shampooed, rinsed again and dried. - 100 cc. The hair is dyed medium ash-blond. Naturally white or nearly white live hair is impregnated with this solution. After contact Example V. 115 for 20 minutes, the excess solution is blotted off, for example by means of a towel, and an Mearly white live hair is impregnated with an impregnating solution as defined in Example I. After a period of contact of 20 aqueous solution having the following composition is then applied. minutes, the excess solution is blotted off and an aqueous solution having the following 120 Sodium chloride -100 cc. composition is applied:-Water 20% ammonia Sodium chloride -2.8 g. 20% ammonia -After contact for 20 minutes, the hair is 6 cc. rinsed clear with water and dried. The hair Hydrogen peroxide (20 vol.) 125 (6% of H<sub>2</sub>O<sub>2</sub>) is dyed a dark grey which gradually changes to black. (This constitutes a progressive dye-Water to make -~ 100 cc. After contact for 20 minutes, the hair is **65** ing).

	rinsed, shampooed, rinsed again and dried. A medium brown shade is obtained.	pooed, rinsed again and dried. A dark grey shade is obtained.
	Example VI.	EXAMPLE X.
_	Naturally white or nearly white hair is	A solution having the following composition
5	impregnated with an impregnating solution as defined in Example I.	is prepared: 5.6-dihydroxy-indole 1 g.
	The solution is left in contact for 20	5,6-dihydroxy-indole 1 g. Water 100 cc.
	minutes, whereafter the excess solution is blotted off and an aqueous solution having	The pH-value of such a solution is 6.2. This solution is applied to naturally white
10	the following composition is applied:—	live hair and left in contact therewith for 15
	Sodium chloride 3.2 g. 20% ammonia 2 cc.	minutes. After blotting off the excess solu-
	Hydrogen peroxide (20 vol.)	tion, an aqueous 4% by weight sodium iodate solution is applied to the hair. After this solu-
_	$(6\% \text{ of } H_0O_0) 16 \text{ cc.}$	tion has been allowed to act for 15 minutes,
15	Water to make 100 cc. This solution is left in contact with the hair	the hair is rinsed, shampooed, rinsed again and
	for 20 minutes, the hair is then rinsed, sham-	dried. The hair is dyed a dark grey, which is not, however, as dark as the shade obtained
	pooed, rinsed again and dried. A light brown	in Example IX.
	shade is obtained.	Example XI.
20	Example VII.	Naturally white live hair is treated with an
	Naturally white live hair is treated with an impregnating solution as defined in Example I.	impregnating solution as defined in Example IX.
	This solution is left in contact with the	After contact for 15 minutes, the excess
25	hair for 30 minutes, whereafter the excess solution is blotted off and an aqueous solution	solution is blotted off and an aqueous 4% by weight sodium periodate solution is then
	having the following composition is applied to	applied to the hair. After contact for 15
	the hair. Sodium chloride 5 g.	minutes, rinsing, shampooing, rinsing again
	Triethanolamine 26 cc.	and drying, the hair is dyed in a grey shade which is a little lighter than that obtained
30	Hydrogen peroxide (20 vol.)	in Example X.
	$(6\% \text{ of } \hat{H}_2O_2)$ 10 cc. Water 75 cc.	EXAMPLE XII.  A solution having the following composition
	After contact for 30 minutes, rinsing, sham-	is prepared:—
35	pooing, rinsing again and drying, the hair is dyed grey.	5,6-dihydroxy-indole 1 g. Water 100 cc.
	Example VIII.	This solution is applied to naturally white
	Almost white live hair is impregnated with an impregnating solution as defined in	live hair and, after 15 minutes, the excess liquid is blotted off and an aqueous 4%
	Example I, which is left in contact therewith	by weight sodium periodate solution is then
40	for 20 minutes. After blotting off the excess	applied to the hair. After contact for 15
	solution, the following solution is applied:— Sodium chloride 4.2 g.	minutes, the hair is rinsed, shampcoed, rinsed again and dried. The hair is dyed a light
	Chloramine T (sodium N-	grey.
45	chloro-p-toluene sulphon- amide) 1.2 g.	EXAMPLE XIII.  Hair is impregnated with an impregnating
	20% ammonia 14 cc.	solution as defined in Example IX.
	Water to make 100 cc.  After contact for 20 minutes, the hair is	This solution is left in contact with the hair for 15minutes, whereafter the excess
	rinsed, shampooed, rinsed again and dried.	liquid is blotted off and an equal volume of
50	The hair is dyed a fine deep black.	an aqueous 5% by weight sodium persulphate solution is applied thereto. After 15 minutes,
	Example IX.	the hair is rinsed, shampooed, rinsed again
	Naturally white live hair is impregnated	and dried. A medium grey is obtained.
	with the following impregnating solution:— 5,6-dihydroxy-indole - 1 g.	EXAMPLE XIV.  An aqueous solution containing 1% by
55	Acetic acid 0.5 cc.	weight of 5,6-dihydroxy-indole is applied to
	Water to make 100 cc. The pH-value of this solution is 3.3. This	naturally white live hair. This solution is left in contact with the hair for 15 minutes.
	solution is left in contact with the hair for	The excess solution is then blotted off and
K۸	15 minutes. The excess solution is then blotted	an aqueous 5% by weight sodium persulphate
60	off and an aqueous 4% by weight of sodium iodate solution is then applied to the hair.	solution is applied. After 15 minutes, the hair is rinsed, shampooed, rinsed again and dried.
	After 15 minutes, the hair is rinsed, sham-	The shade obtained is medium grey.

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	EXAMPLE XV.  Naturally white live hair is impregnated with an impregnating solution as defined in	It is to be understood that the invention includes dead hair or similar keratinous material dyed by employing the dying pro-	65
5	Example I.  When this solution has been allowed to act	cess according to the application e.g., hair, bristles, felt, fur and like keratinous products.	
	for 20 minutes, the excess solution is blotted off and a solution having the following composition is applied:—	WHAT WE CLAIM IS:—  1. A process for dyeing hair and like kera-	70
10	20% ammonia 15 cc. Sodium iodate 0.5 g.	tinous materials, including live hair, which consists essentially in impregnating the hair or like keratinous material at room tempera-	
	Water to make 100 cc. After contact for 20 minutes, the hair is rinsed, shampooed, rinsed again and dried. There is obtained a very dark grey shade hav-	ture with an aqueous solution of 5,6-dihyd- roxy-indole having a pH-value of at most 7, allowing the impregnation to proceed for 5	75
15	ing a violet tinge, gradually changing to black.	to 60 minutes, then blotting off the excess solution and applying to the hair or like	
	EXAMPLE XVI.  Almost white live hair is impregnated with an impregnating solution as defined in	keratinous material an aqueous solution capable of inducing oxidation of the 5,6-dihydroxy-indole.	80
20	Example I which is left in contact therewith for 20 minutes. The excess solution is then blotted off and a solution having the following	2. A process according to Claim 1 wherein said impregnating solution contains acetic acid or monochloroacetic acid.	
	composition is applied thereto:— Sodium periodate 0.5 g.	3. A process according to Claim 1 or 2 wherein the concentration of the 5,6-dihyd-	85
25	20% ammonia 15 cc.  Water to make 100 cc.  This solution is allowed to act for 20	roxy-indole in the impregnating solution is 0.1 to 10% by weight.  4. A process according to Claim 3 where-	00
	minutes and the hair is then rinsed, sham- pooed, rinsed again, and dried. There is	in said concentration is 1 to 2% by weight.  5. A process according to any of Claims 1 to 4 wherein the aqueous solution capable	90
30	directly obtained a fine black shade, which is substantially permanent.  EXAMPLE XVII.	of inducing oxidation is one which induces aerial oxidation.	
	Live, almost white hair is impregnated with an impregnating solution as defined in	6. A process according to Claim 5 wherein said solution contains ammonia, a basic	95
35	After contact for 20 minutes, the excess liquid is blotted off and a solution having the following composition is applied to the hair:—	ammonium salt or an organic amine.  7. A process according to Claim 5 or 6 wherein said solution also contains an oxidation catalyst.	100
40	Ammonium persulphate 0.5 g. 20% ammonia 15 cc. Water to make 100 cc.	8. A process according to any of Claims 1 to 4 wherein the solution applied to induce oxidation is one containing an oxidising agent.	
	This solution is left in contact with the hair for 20 minutes. After rinsing, shampooing rinsing again and drying, the hair is dyed a	9. A process according to Claim 8 wherein the oxidising agent is hydrogen peroxide, an alkali metal or ammonium iodate, periodate or persulphate or an alkali metal salt of an	105
45	fine deep black.  EXAMPLE XVIII.  A solution having the following composition	N-chlorinated derivative of a sulphonamide.  10. A process according to Claim 8 where-	110
50	is employed to impregnate almost white live hair:— 5,6-dihydroxy-indole 1 g.	is one containing hydrogen peroxide, an alkali metal or ammonium iodate or periodate or an	
50	Monochloroacetic acid - 1 g.  Water to make 100 cc.  After contact for 20 minutes with the hair,	alkali metal salt of an N-chlorinated deriva- tive of a sulphonamide and is made alkaline. 11. A process according to Claim 8 where-	115
55	Ammonium persulphate 6 g.	is one containing an alkali metal or ammonium persulphate and is made acid.  12. A process according to Claim 1 when	
۷۸	Sodium chloride 4 g.  Monochloroacetic acid 3 g.  Water to make 100 cc.  This solution is left in contact with the air	carried out substantially as set forth in any one of the foregoing specific Examples I to XVIII.	120
60	for 20 minutes, whereafter the hair is rinsed, shampooed, rinsed again and dried. The hair is dyed a medium grey.	13. Dead hair or like keratinous materials	125

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